

Selective Azetidine and Tetrahydropyridine Formation via Pd-Catalyzed Cyclizations of Allene-Substituted Amines and Amino Acids

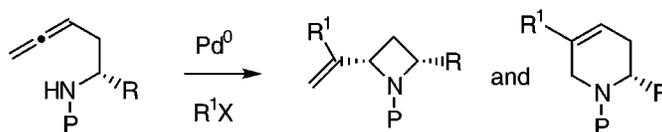
Floris P. J. T. Rutjes,* Kim C. M. F. Tjen, Larissa B. Wolf, Willem F. J. Karstens, Hans E. Schoemaker,[†] and Henk Hiemstra*

Institute of Molecular Chemistry, University of Amsterdam, Nieuwe Achtergracht 129, 1018 WS Amsterdam, The Netherlands, and DSM Research, Department of Organic Chemistry and Biotechnology, P.O. Box 18, 6160 MD Geleen, The Netherlands

floristr@org.chem.uva.nl

Received June 2, 1999

ABSTRACT



R = H, CO₂Me; R¹ = aryl, alkenyl

By choosing the right substituents either highly functionalized unusual four-membered ring amino acids or the isomeric pipercolic acid derivatives are obtained in enantiomerically pure form. Starting material is a linear allene-containing amino acid that has been resolved via biocatalysis.

Heteroannulation processes involving unsaturated functionality and Pd catalysis have been extensively studied over the past decade.¹ Beside the use of olefins and acetylenes as the

electrophilic partners, allenes in particular have attracted the attention of organic chemists in recent years.^{2,3} In all of these examples the heteronucleophile attacks one of the sp²-carbon atoms of the allene.⁴ Recently, unprecedented alternative behavior was reported by our group, where a lactam nitrogen atom—with a two-carbon tether between the allene and the nitrogen atom—reacted at the sp carbon atom of the allene to form five-membered ring enamides.^{5,6} In conjunction with this work and similar Pd-catalyzed reactions of acetylenic amino acids in our group,⁷ we wish to present cyclizations

[†] DSM Research.

(1) (a) Tsuji, J. *Palladium Reagents and Catalysts*; Wiley: New York, 1995. (b) For a general review on metal-catalyzed amination, see: Müller, T. E.; Beller, M. *Chem. Rev.* **1998**, *98*, 675.

(2) For nitrogen nucleophiles, see for example: (a) Larock, R. C.; Tu, C.; Pace, P. *J. Org. Chem.* **1998**, *63*, 6859. (b) Desarbre, E.; Mérour, J. Y. *Tetrahedron Lett.* **1996**, *37*, 43. (c) Larock, R. C.; He, Y.; Leong, W. W.; Han, X.; Refvik, M. D.; Zenner, J. M. *J. Org. Chem.* **1998**, *63*, 2154. (d) Larock, R. C.; Zenner, J. M. *J. Org. Chem.* **1995**, *60*, 482. (e) Kimura, M.; Tanaka, S.; Tamaru, Y. *J. Org. Chem.* **1995**, *60*, 3764. (f) Davies, I. W.; Scopes, D. I. C.; Gallagher, T. *Synlett* **1993**, 85. (g) Kimura, M.; Fugami, K.; Tanaka, S.; Tamaru, Y. *J. Org. Chem.* **1992**, *60*, 6377. (h) Prasad, J. S.; Liebeskind, L. S. *Tetrahedron Lett.* **1988**, *29*, 4257. (i) Shimizu, I.; Tsuji, J. *Chem. Lett.* **1984**, 233.

(3) For oxygen nucleophiles, see for example: (a) Ma, S.; Shi, Z. *J. Org. Chem.* **1998**, *63*, 6387. (b) Jonasson, C.; Bäckvall, J. E. *Tetrahedron Lett.* **1998**, *39*, 3601. (c) Walkup, R. D.; Guan, L.; Mosher, M. D.; Kim, S. W.; Kim, Y. S. *Synlett* **1993**, 88. (d) Walkup, R. D.; Guan, L.; Kim, Y. S.; Kim, S. W. *Tetrahedron Lett.* **1995**, *36*, 3805. (e) Besson, L.; Bazin, J.; Goré, J.; Cazes, B. *Tetrahedron Lett.* **1994**, *35*, 2881.

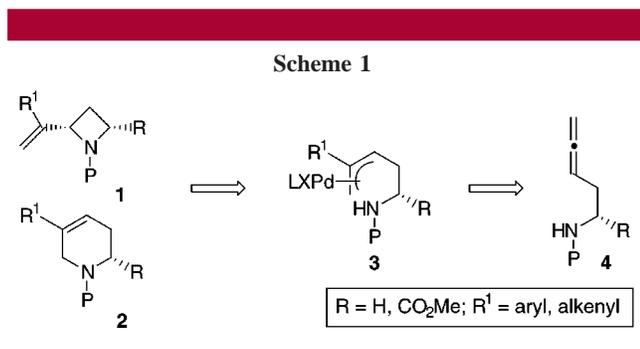
(4) Reference 1, p 166.

(5) (a) Karstens, W. F. J.; Rutjes, F. P. J. T.; Hiemstra, H. *Tetrahedron Lett.* **1997**, *38*, 6275. (b) Karstens, W. F. J.; Stol, M.; Rutjes, F. P. J. T.; Hiemstra, H. *Synlett* **1998**, 1126.

(6) More recently, similar regiochemistry was observed in lanthanide-catalyzed cyclizations of aminoallenes: Arredondo, V. M.; McDonald, F. E.; Marks, T. *J. Am. Chem. Soc.* **1998**, *120*, 4871.

(7) Wolf, L. B.; Tjen, K. C. M. F.; Rutjes, F. P. J. T.; Hiemstra, H.; Schoemaker, H. E. *Tetrahedron Lett.* **1998**, *39*, 5081.

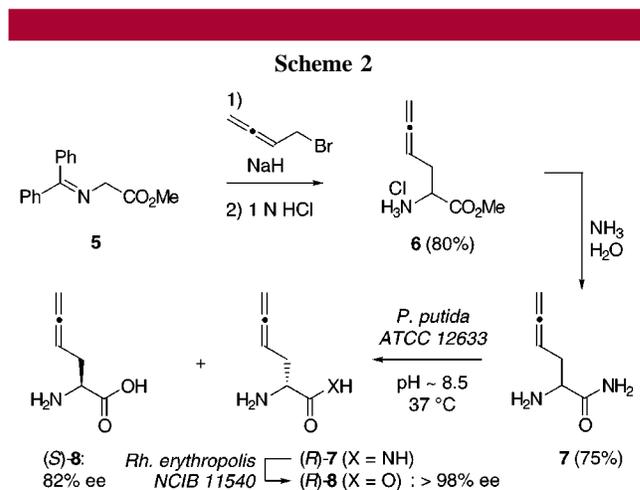
of the linear β -aminoallenes **4** ($R = H, CO_2Me$; $P =$ protecting group), where the allene and the amine are also separated by a two-carbon tether (Scheme 1).



In this case, four- and six-membered ring amines and amino acids are obtained with high selectivity.⁸ Interestingly, attempts of the group of Yamamoto to cyclize such an unsubstituted β -aminoallene (viz., **4**, $R = H$, $P = Ts$) failed, resulting in the isomerized diene product.⁹ Very recently, the group of Kang presented a single example of the cyclization of this unsubstituted aminoallene, where in a moderate yield both the four- and six-membered heterocycle were obtained in a 2:1 ratio. The latter example, however, is rather limited in terms of selectivity, substitution pattern, and yield.¹⁰ Our method, on the other hand, holds—in addition to the selective formation of four- and six-membered rings—a promising potential for introduction of various R^1 substituents and thus arrive at unique enantiopure amino acid derivatives.

The starting aminoallenes were synthesized as follows. The unsubstituted β -aminoallene **4** ($R = P = H$) was obtained via a literature procedure, involving a Crabbé reaction of 3-butyn-1-ol and conversion of the resulting alcohol into the

amine.¹¹ The enantiomerically pure allene-containing amino acid **8** was prepared via an enzymatic resolution of the corresponding racemic amide **7** (Scheme 2).¹²



The latter compound was synthesized via (i) alkylation of the glycine-derived ketimine **5**¹³ with 4-bromo-1,2-butadiene (prepared from 2,3-butadienol¹⁴ with Br_2PPh_3 and imidazole), (ii) acid hydrolysis of the ketimine to give **6**, and (iii) reaction with aqueous ammonia affording the amino acid amide **7**.¹⁵ Subjection to the enzymatic resolution conditions (aminopeptidase produced by *Pseudomonas putida* ATCC 12633,¹⁶ pH 8.5, 37 °C, 60 h)¹⁷ and separation of the resulting acid and amide¹⁵ provided the (*S*)-acid **8** in 44% yield and 82% ee after purification by ion exchange chromatography.¹⁸ The (*R*)-amide **7** was hydrolyzed by subjecting to a nonspecific amidase produced by *Rhodococcus erythropolis* NCIB 11540¹⁹ to afford (*R*)-**8** in 35% yield (two steps) and $> 98\% ee$.²⁰

Functionalization to the cyclization precursors proceeded by using standard methodology leading to the desired precursors **9–10** and **21–24** in good yields and without detectable racemization. The cyclization results for the unsubstituted aminoallenes **9** and **10** are shown in Table 1. Application of the cyclization conditions (10 mol % Pd(PPh_3)₄, 5 equiv of K_2CO_3 , and 5 equiv of R^1X , DMF, 80 °C)^{2f} onto allene **9** (entry 1) led in 30 min to complete conversion, affording a 67:33 ratio of the four- and six-membered rings **16a** and **b**, respectively, which is in line with the result of Kang.¹⁰ Upon prolonged reaction times, this ratio changed in favor of the six-membered ring (entry 2).²¹ Use of the pyridine-derived iodide **13** gave a similar

(8) Similar three- vs five-membered ring formation of α -amino allenes was reported recently: Ohno, H.; Toda, A.; Miwa, Y.; Taga, T.; Osawa, E.; Yamaoka, Y.; Fuji, N.; Ibuka, T. *J. Org. Chem.* **1999**, *64*, 2992.

(9) Meguro, M.; Yamamoto, Y. *J. Org. Chem.* **1998**, *39*, 5421.

(10) Kang, S.-K.; Baik, T.-G.; Kulak, A. N. *Synlett* **1999**, 324.

(11) (a) Li, D.; Zhou, H.-Q.; Dakoji, S.; Shin, I.; Oh, E.; Liu, H.-W. *J. Am. Chem. Soc.* **1998**, *120*, 2008. (b) Hanack, M.; Haefner, J. *Chem. Ber.* **1966**, *99*, 1077.

(12) 2-Amino-4,5-hexadienoic acid (homo-allenylglycine) is a naturally occurring amino acid of which several syntheses have been published. Isolation: Chilton, W. S.; Tsou, G.; Kirk, L.; Benedict, R. G. *Tetrahedron Lett.* **1968**, 6283. Racemic synthesis: (b) Cazes, B.; Djahanbini, D.; Goré, J.; Genêt, J.-P.; Gaudin, J.-M. *Synthesis* **1988**, 983. (c) Black, D. K.; Landor, S. R. *J. Chem. Soc. (C)* **1968**, 281. (d) Black, D. K.; Landor, S. R. *J. Chem. Soc. (C)* **1968**, 283. (S)-Enantiomer: (e) Baldwin, J. E.; Adlington, R. M.; Basak, A. J. *Chem. Soc., Chem Commun.* **1984**, 1284. (S)-Enantiomer (benzyl ester): (f) Dunn, M. J.; Jackson, R. F. W.; Pietruszka, J.; Turner, D. *J. Org. Chem.* **1995**, *60*, 2210.

(13) (a) O'Donnell, M. J.; Polt, R. L. *J. Org. Chem.* **1982**, *47*, 2663. (b) O'Donnell, M. J.; Wojciechowski, K.; Ghosez, L.; Navarro, M.; Sainte, F.; Antoine, J.-P. *Synthesis* **1984**, 313.

(14) Bailey, W. J.; Pfeifer, C. R. *J. Org. Chem.* **1955**, *20*, 1337.

(15) The amide was separated from residual acid via formation of the Schiff base ($PhCHO$, H_2O , pH ~ 9), extraction from the water layer, and precipitation upon hydrolysis in acetone using 1 equiv of HCl.

(16) The aminopeptidase was generously provided by DSM Research.

(17) For related examples, see: (a) Schoemaker, H. E.; Boesten, W. H. J.; Broxterman, Q. B.; Roos, E. C.; Kaptein, B.; van den Tweel, W. J. J.; Kamphuis, J.; Rutjes, F. P. J. T. *Chimia* **1997**, *51*, 308–311. (b) Schoemaker, H. E.; Boesten, W. H. J.; Kaptein, B.; Roos, E. C.; Broxterman, Q. B.; van den Tweel, W. J. J.; Kamphuis, J. *Acta Chem. Scand.* **1996**, *50*, 225–233.

(18) The ee was determined using chiral HPLC (Crownpak CR(+)) following a known protocol: Miyazawa, T.; Iwanaga, H.; Yamada, T.; Kuwata, S. *Chem. Express* **1991**, 6, 887.

(19) Boesten, W. H. J.; Cals, M. J. H. U.S. Patent 4 705 752, 1987; *Chem. Abstr.* **1987**, *105*, 170617k.

(20) By using an aminopeptidase produced by a genetically modified *Escherichia coli* strain both enantiomers were produced in a single run in $> 98\% ee$. This experiment, however, has not yet been carried out on a preparative scale. Sonke, T.; Boesten, W. H. J.; Broxterman, Q. B.; Kamphuis, J.; Formaggio, F.; Toniolo, C.; Rutjes, F. P. J. T.; Schoemaker, H. E. In *Stereoselective Biocatalysis Handbook*; Patel, R. N., Ed.; Marcel Dekker, in press.

Table 1

entry	substrate	R ¹ X	time	ratio (a:b) ^a	product (yield) ^b
1	9 (P = Ts)	PhI	0.5 h	67:33	16 (67%)
2	9	PhI	40 h	10:90	16 (34%)
3	9	13	4 h	25:75	17 (55%)
4	9	14	10 min	100:0	18 (34%)
5	9	15	15 min	95:5	19 (41%)
6	10	PhI	4 h	0:100	20 (55%)

^aDetermined by analysis of the ¹H NMR data of the crude mixture. ^bCombined isolated yield after column chromatography.

result, with the six-membered ring being the major isomer (entry 3). In contrast, reactions with the vinyl triflates **14** and **15** led to pleasing results. The azetidine **18a** was formed selectively by using triflate **14** (entry 4), while reaction with triflate **15** gave a selectivity of 95:5 in favor of the four-membered ring **19a** (entry 5). On the other hand, subjection of aminoallene **10**—protected with the more easily removable *p*-nitrobenzenesulfonyl (Ns) group—to these reaction conditions provided the tetrahydropyridine **20b** selectively in 55% yield (entry 6).

To study the influence of the ester substituent, the enantiopure amino acid derivatives **21–24** were subjected to identical cyclization conditions (Table 2). Initially, **21** was

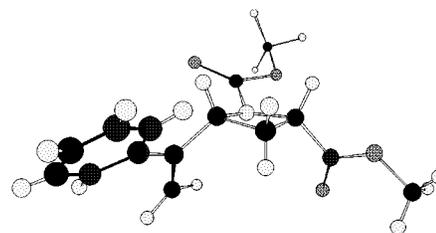
Table 2

entry	allene	R ¹ X	solvent	T (°C)	time	ratio (a:b) ^a	product (yield) ^b
1	21 (P = Ts)	PhI	DMF	80	4 h	0:100	25 (78%)
2	21	PhI	DMF	rt	4 h	50:50	25 (80%)
3	21	PhI	DMF	80	10 min	70:30	25 (84%)
4	21	PhI	THF	60	1.5 h	88:12	25 (45%)
5	21	14	DMF	80	15 min	88:12	26 (96%)
6	21	15	DMF	80	10 min	95:5	27 (77%)
7	22 (P = Bn)	PhI	DMF	70	3 h	0:100	28 (76%)
8	23 (P = CO ₂ Me)	PhI	DMF	80	16 h	42:58	29 (33%)
9	24 (P = Ac)	PhI	DMF	80	2 h	67:33	30 (12%)

^aDetermined by analysis of the ¹H NMR data of the crude mixture. ^bCombined isolated yield of **20** and **21** after flash chromatography. In all entries, both products were obtained without loss of enantiopurity according to chiral HPLC (Chiralcel OD).

treated under similar conditions as **9** for 4 h (entry 1), resulting selectively in the pipercolic acid derivative **25b** in 78% yield. Closer examination of the conditions revealed that with shorter reaction times a considerable amount of the four-membered ring **25a** was formed as a single *cis*-isomer.

The best example is shown in entry 3: the reaction was finished in 10 min, with **25a** being the major product. Surprisingly, the reaction also proceeded at room temperature, but did not show any selectivity (entry 2). To improve the ratio in favor of the four-membered ring, different parameters (temperature, solvent, and reaction time) were varied, resulting in a maximum selectivity of 88:12 (THF, 60 °C), albeit the yield was moderate (entry 4). Again, interesting results were obtained by using the vinyl triflates **14** and **15**. This led to fast reactions (finished in 10 min) and excellent yields of the four-membered amino acids **26a** and **27a**, with only a small degree of formation of the six-membered rings (entries 5 and 6). The benzylated precursor **22** led in a clean reaction to the pipercolic ester **28b** (entry 7). This result is in accord with previous reports on cyclizations of amines onto π -allylpalladium intermediates, where also complete conversion into the thermodynamic product was encountered.²² Introduction of a methyl carbamate (entry 8) or an amide function (entry 9) did not lead to satisfactory results: the low yields did not encourage further investigations into optimization of the product ratio. Interestingly, the carbamate-functionalized azetidine ester **29a** was the only product that gave crystals that could be subjected to an X-ray structure determination. Thus, the structure and absolute orientation of the substituents in the azetidine were unambiguously established (Figure 1).²³

Figure 1. Crystal structure of **29a**.

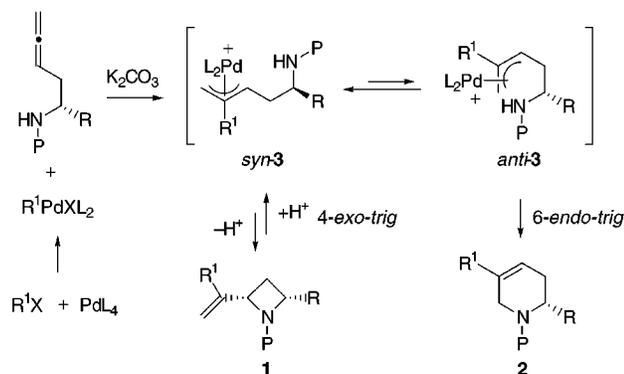
An explanation for these results can be found by considering the different pathways that play a role (Scheme 3). Initially, reaction of the allene with the in situ formed

(21) A similar ring expansion (from three- to five-membered rings) has been reported previously: Fugami, K.; Morizawa, Y.; Oshima, K.; Nozaki, H. *Tetrahedron Lett.* **1985**, 26, 857.

(22) Review: Trost, B. M. *Angew. Chem., Int. Ed. Engl.* **1989**, 28, 1173.

(23) X-ray data for **29a**: C₁₅H₁₇NO; *M*_r = 227.3; monoclinic; *C*₂; *a* = 12.994(2), *b* = 9.907(1), *c* = 10.371(2) Å; β = 111.36(1)°; *V* = 1243.4(3) Å³; *Z* = 4, *D*_x = 1.21 g cm⁻³; λ (Cu K α) = 1.5418 Å, μ (Cu K α) = 5.57 cm⁻¹; *F*(000) = 488, -25 °C. Final *R* = 0.040 for 1064 observed reflections. Crystallographic data (excluding structure factors) have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC-114107.

Scheme 3



organopalladium(II) species will give rise to the more stable *syn*- π -allylpalladium complex **3**,²⁴ which via π - σ - π isomerization can be converted to the more hindered *anti*-isomer.²⁵

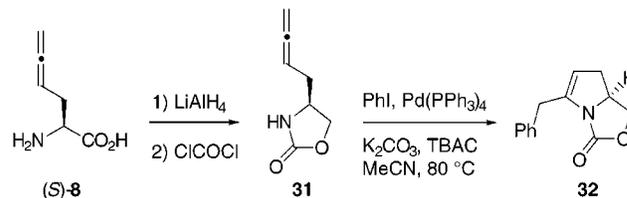
The *syn*-isomer can only cyclize in a 4-*exo-trig*-fashion to give the kinetic product **1**. The four-membered ring, however, can under the influence of Pd(PPh₃)₄ undergo ring opening to regenerate the π -allylpalladium complex, which eventually via the equilibrium with the *anti*-complex under the reaction conditions can isomerize to the thermodynamically more stable six-membered ring **2**.²⁶ The rate of isomerization will depend on the substituents P and R¹. For example, for R¹ = alkenyl, it is more difficult to form the intermediate complex so that isomerization does not rapidly occur and a relatively large amount of the four-membered ring is formed. On the other hand, improving the leaving group ability of the nitrogen (by going from Ts to Ns) enhances the isomerization process in such a way that the four-membered ring is no longer observed.

Thus, by changing different parameters either four- or six-membered ring formation is observed resulting from attack of the nitrogen onto one of the sp²-allene carbon atoms. Remarkably, a slight substrate modification—ester reduction followed by oxazolidinone formation (viz., **31**, Scheme 4)—

(24) An alternative cyclization mechanism—activation of the allene by the organopalladium(II) species as the π -complex followed by nucleophilic attack of the nitrogen, which has been proposed by Walkup (ref 3c,d) and Gallagher (ref 2f), cannot be ruled out.

results in attack of the nitrogen on the central sp allene carbon atom under similar reaction conditions, providing the corresponding five-membered ring **32**.^{5b}

Scheme 4



In conclusion, we have shown that Pd-catalyzed cyclizations of β -aminoallenes can selectively lead to four- or six-membered nitrogen heterocycles without loss of optical activity in high yields. In particular vinyl triflates show a promising selectivity in the cyclization processes leading to the four-membered rings. In addition, we have shown that via conversion into the more restricted oxazolidinone analogues, the corresponding five-membered rings are also accessible in both the (*R*)- and (*S*)-conformation. The mechanistic basis of this remarkable behavior is currently under further investigation.

Acknowledgment. DSM Research is gratefully acknowledged for providing research grants to K.C.M.F.T. and L.B.W. This research has been financially supported by the Council for Chemical Sciences of The Netherlands Organization for Scientific Research (CW-NWO). J. Fraanje and K. Goubitz (Laboratory of Crystallography, University of Amsterdam) are acknowledged for the X-ray structure determination.

Supporting Information Available: A general procedure for the cyclization reactions and full characterization for compounds **7**, **8**, **18a**, **20b**, **25a,b**, **26a**, and **27a**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL990700C

(25) For a similar discussion, see: Ahmar, M.; Cazes, B.; Goré, J. *Tetrahedron Lett.* **1985**, 26, 3795.

(26) The four- to six-membered ring isomerization was proven in a separate experiment via subjection of **25a** to the cyclization conditions, resulting in complete conversion into the tetrahydropyridine **25b**.